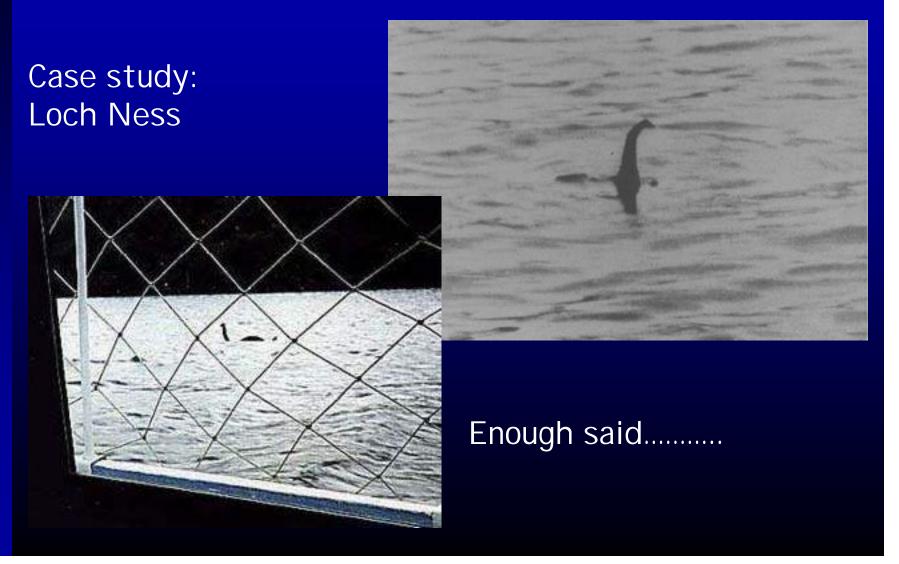




Invasive species What's the problem?



What is ozone?

- Allotropic form of oxygen (O3)
- Colorless gas
- Characteristic odor
- Extremely reactive oxidizing agent

Ozone use in aquatic systems

- Disinfection
- Algal control
- Removal/reduction of odor and color causing organics
- Turbidity reduction
- Oxidation of nitrites
- Increased biodegradability of organics
- Improved coagulation of particulate matter

Primary concerns

- How much ozone is required for an effective treatment?
 - Concentration
 - Exposure/contact time
 - Exposure frequency
 - Species/life stage/habitat specificity
- What residual toxicity needs to be mitigated prior to discharge?
 - Whole effluent toxicity (WET) tests

Ozone chemistry in seawater

- Reaction products
 - Ozone
 - "Reactive" products
 - Hypobromous acid (HOBr)
 - Hypobromite ion (OBr-)
 - "Stable" products
 - Bromoform (CHBr₃)
 - Bromate (BrO₃-)
- Measurement
 - Total Residual Oxidants (TRO)
 - Oxidation:Reduction Potential (ORP)

Two phased research approach

Laboratory



Field



Laboratory testing program

 Determine the relationship between the exposure concentration and the time of exposure

Determine a range of species

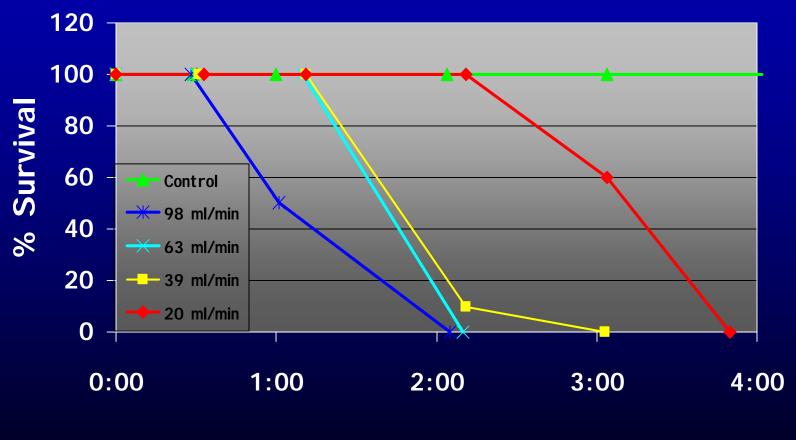
sensitivity

- Bacteria → fish

Laboratory testing program

- Determine factors that may effect exposure concentrations
 - Flocculent additives
 - Salinity
 - Organic carbon load
 - Suspended solids
 - Temperature

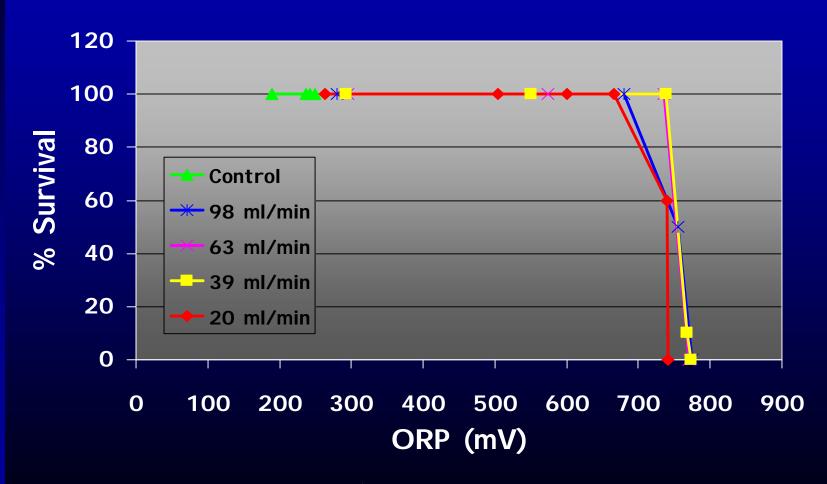
Sheepshead minnow survival vs. time



Time (hrs)

 O_3 concentration: 0.07 mg O_3 /ml

Sheepshead minnow survival vs. ORP concentration



 O_3 concentration: 0.07 mg O_3 /ml



Ozone generation system

Generates 1,800
g/O₃/h

 Distributed through a series of ceramic diffusers in each ballast tank



- Evaluate the efficacy of O₃ as a control agent under field conditions
 - Exposure
 - Concentration of O₃ products
 - Spatial and temporal heterogeneity
 - Effects
 - Examine the toxicity of O₃ to a range of organisms under field conditions

- Exposure measures
 - Niskin bottle samples
 - TRO/ORP/Bromate/ Bromoform concentrations
 - Examine spatial heterogeneity



- Effects measures
 - Bacterial counts
 - Plankton tows

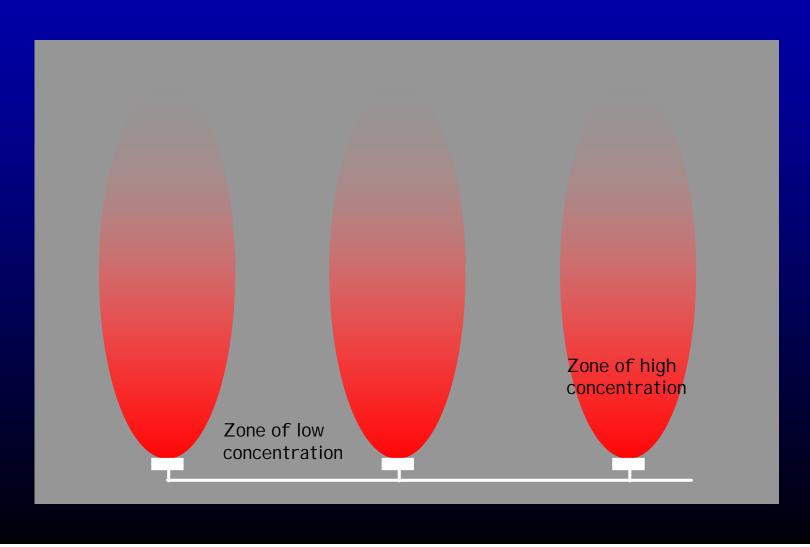


- Effects measures
 - In situ toxicity tests
 - Benthic sediment
 - Rhepoxynius abronius
 - Water-column cages
 - Mysid shrimp
 - Sheepshead minnows
 - Shore crabs



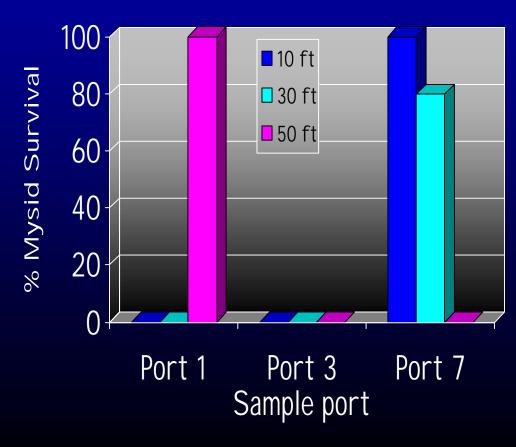


Exposure is an important consideration



Organism survival varied with species and tank position following 10hr O₃ exposure

- Plankton & bacteria showed high mortality
- Mysid survival varied with depth
- Sheepshead minnows died at all depths
- Shore crabs and benthic invertebrates unaffected



WET testing

- Effluent discharges require testing to ensure no adverse effects on organisms in the receiving waters
 - Chemical specific parameters
 - WET testing
 - Acute
 - Mysid shrimp
 - Topsmelt
 - Chronic
 - Echinoderm or mollusk embryo/larval



Conclusions

- Evaluation of any ballast water treatment technology should include both laboratory tests and field validation
- Evaluation of chemical control agents must include an assessment of exposure concentration and duration, efficacy, and residual toxicity

Conclusions

- Ozone is an effective agent for the control of a variety of aquatic organisms in ballast waters although there is variability due to species sensitivity and O₃ distribution
- Post-ozonation residual toxicity of ballast waters is low